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(b) according to mendeleev's original design. (c) according to increasing atomic number. (d) based on when they were discovered. **2.2 periodic law 2.2 the periodic law - 2.2 periodic law page 3 april 24, 2001 discrepancies disappeared when the elements were arranged by this feature rather than atomic mass. moseley's work led the conclusion that the chemical and physical properties example exercise 6.1 periodic law - start here. get there.** - example exercise 6.1 periodic law. mendeleev proposed that elements be arranged according to increasing atomic mass. beginning with rb, each of the elements in the fifth row increases in atomic mass until iodine. although the atomic numbers of te (52) and i (53) increase, the atomic masses of te (127.60) and i (126.90) do not. experimentally ... **laboratory manual - student edition - glencoe** - laboratory manual student edition. hands-on learning: laboratory manual, se/te forensics laboratory manual, se/te ... chapter 6 the periodic table and periodic law ... never work alone in the lab-oratory. work only when your teacher is present. **periodic trends in reactivity - linusempt.ku** - periodic trends introduction in the modern periodic table (shown below in figure 1), elements are arranged according to in- ... ance with periodic law and can be predicted by periodic trends in first ionization energies, ie 1, (see figure 3) and the placement of these ... and au), 16, 17, and 18 are commonly referred to as the alkaline earth ... **the periodic law lab answers - cities.expressindia** - periodic law lab answer key bing riverside resortnet indexed periodic law lab 17 answer key pdf or periodic law lab 17 answer key pdf data that are online ch 5 lab periodic law chema23 purpose in this experiment you will use your knowledge of periodic properties and a list of clues to correctly arrange the elements from a scrambled periodic ...

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